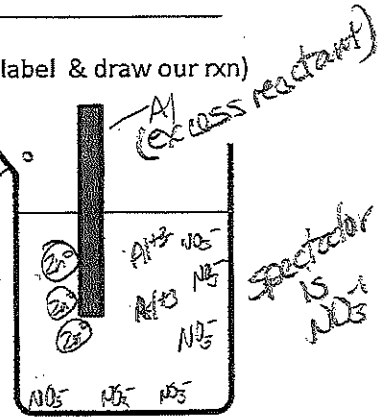
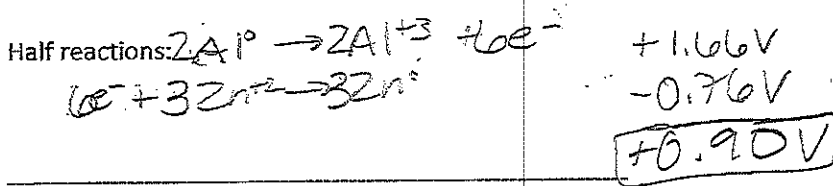
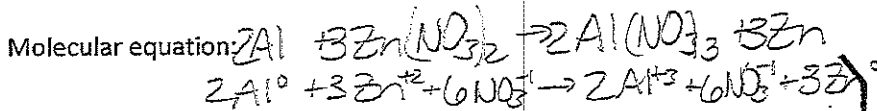
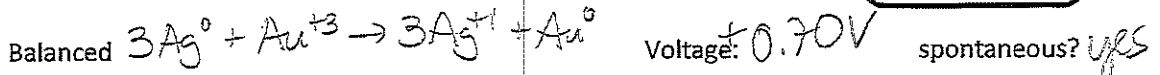
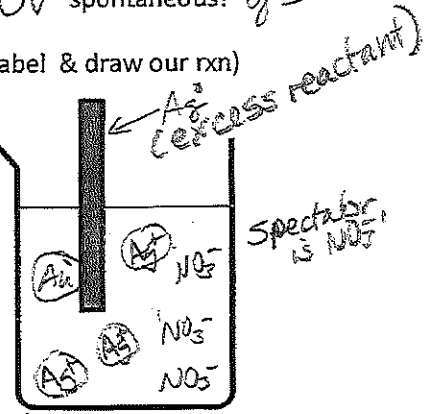
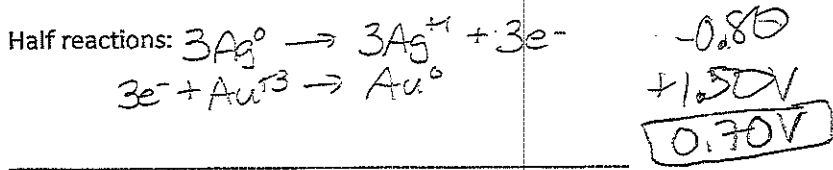
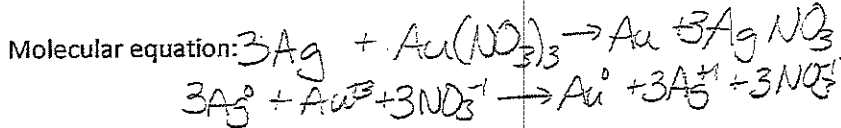


1. Aluminum metal is immersed into an aqueous solution of Zinc nitrate. (label & draw our rxn)

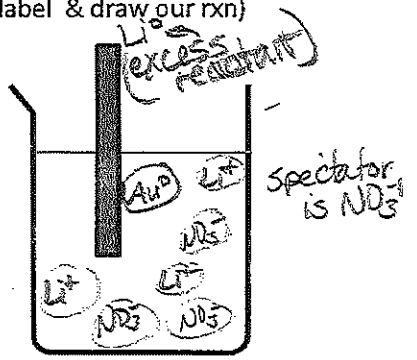
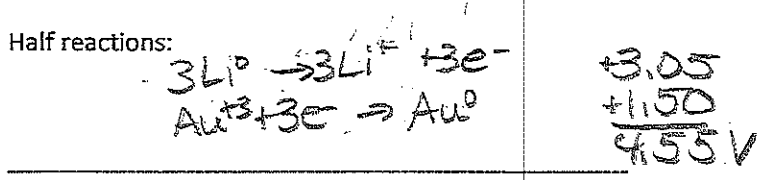
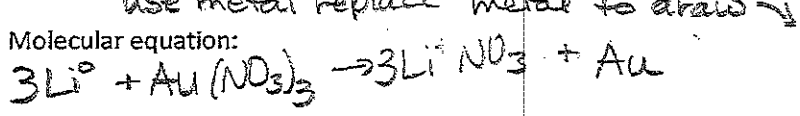


2. Silver metal is immersed into an aqueous solution of Gold(III) nitrate. (label & draw our rxn)



3. Create a reaction that would produce the <sup>high</sup> most voltage possible. (label & draw our rxn)

*example use Au + Li*



*clings to excess Li*